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Formal total synthesis of hemibrevetoxin B by a convergent strategy

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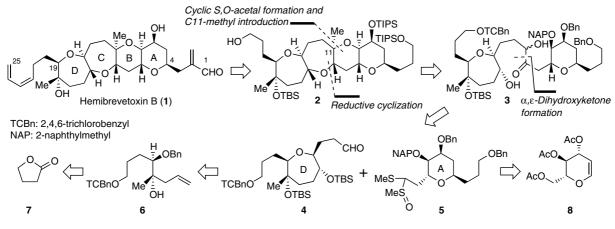
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Abstract—Concise construction of the *trans*-fused 7/7/6/6 tetracyclic ether part of hemibrevetoxin B (1) was achieved by a convergent strategy based on coupling reaction of an acyl anion equivalent, reductive cyclization of an α,ϵ -dihydroxyketone, and introduction of a methyl group at the central ring junction by the Nicolaou method. The resultant tetracyclic ether was transformed into the known intermediate, which was already converted to 1 by the Yamamoto group, thereby completing the formal total synthesis of 1.

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Hemibrevetoxin B (1), isolated as a congener of brevetoxins from dinoflagellate *Gymnodinium breve* by Prasad and Shimizu in 1989,¹ has a *trans*-fused 7/7/6/6 tetracyclic ether structure. So far, many research groups have finished the total synthesis of $1.^2$ Notably, all of them, except the Holton group,^{2k} adopted a linear synthetic strategy, though a convergent strategy would have comparable efficiency even in the synthesis of such small polyether. In this context, synthesis of 1 through a convergent strategy, which would allow the large-scale preparation of 1, was planned. Here, a concise formal total synthesis of 1 from the A- and D-ring segments (5 and 4, respectively) is described.

Our synthetic plan, illustrated in Scheme 1, focuses on a short-step construction of the central BC-ring part of 1. We envisioned the synthesis of tetracyclic ether 2, which was already converted to 1 by the Yamamoto group,^{2d}



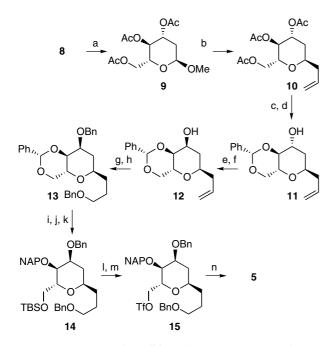
Scheme 1.

Keywords: Polyether; Hemibrevetoxin B; Natural product synthesis; Acyl anion equivalent; Convergent synthesis.

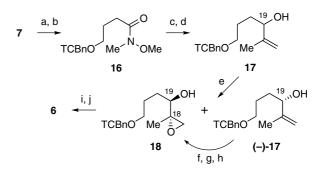
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through reductive cyclization of α, ε -dihydroxy ketone $3^{3,4b,c}$ followed by *S*,*O*-acetal formation and introduction of a methyl group at C11 according to the Nicolaou procedure.⁵ The intermediate **3** would be constructed by a coupling reaction of the A-ring **5** having a dimethyldithioacetal mono-*S*-oxide group as an acyl anion equivalent with the D-ring **4**, and the subsequent deprotection.^{4,6} The D-ring **4** could be synthesized from γ -butyrolactone **7** via **6** and the A-ring **5** could be prepared from tri-*O*-acetyl D-glucal **8**.

Synthesis of **5** is shown in Scheme 2. Transformation of **8** into methyl acetal **9** under the Yadav conditions⁷ followed by allylation gave **10** as a single stereoisomer in 63% yield for two steps. The triacetate **10** was subjected to methanolysis and the subsequent selective protection of the 1,3-diol part to produce **11** in 70% overall yield, in which the free α -hydroxy group was inverted through Swern oxidation⁸ followed by reduction with K-Selectride to afford **12** in 95% yield for two steps. Conversion of **12** to a diol by hydroboration/oxidation and protection of the resulting diol part provided **13** in 55% overall yield. The benzylidene acetal part of **13** was removed to give a diol, which was selectively protected with a TBS group at a primary hydroxy group and with a 2-naphthylmethyl (NAP) group⁹ at a secondary hydroxy group to



Scheme 2. Reagents and conditions: (a) MeOH, NaI, CeCl₃·7H₂O, MeCN, reflux, 3.5 h, 66%; (b) allyltrimethylsilane, TMSOTf, MeCN, $0 \rightarrow 23 \,^{\circ}$ C, 1.5 h, 96%; (c) NaOMe, MeOH, $0 \rightarrow 23 \,^{\circ}$ C, 3 h, 88%; (d) PhCHO, TsOH·H₂O (cat.), PhH–DMF (3:1), reflux, 7 h, 79%; (e) (COCl)₂, DMSO, CH₂Cl₂, -78 $\,^{\circ}$ C, 15 min, then Et₃N, -78 $\rightarrow 0 \,^{\circ}$ C, 30 min, 100%; (f) K-Selectride, THF, -78 $\,^{\circ}$ C, 1 h, 95%; (g) 9-BBN, THF, 23 $\,^{\circ}$ C, 1 h, then TBHP, 3 M NaOH, $0 \rightarrow 23 \,^{\circ}$ C, 40 min; (h) BnBr, 'BuOK, Bu₄NI, THF, 23 $\,^{\circ}$ C, 4 h, 55% from 12; (i) 6 M HCl–THF (1:1), 23 $\,^{\circ}$ C, 10 h, 94%; (j) TBSCl, Et₃N, DMAP, CH₂Cl₂, 23 $\,^{\circ}$ C, 160 min, 99%; (k) NAPBr, NaH, Bu₄NI, THF, $0 \rightarrow 25 \,^{\circ}$ C, 14 h, 100%; (l) TBAF, THF, 25 $\,^{\circ}$ C, 2 h, 99%; (m) Tf₂O, 2,6-lutidine, CH₂Cl₂, -78 $\,^{\circ}$ C, 30 min, 100%; (n) MeSCH₂S(O)Me, BuLi, THF, -78 $\,^{\circ}$ C, 20 min, then 15, $-78 \rightarrow -20 \,^{\circ}$ C, 30 min, 96%.

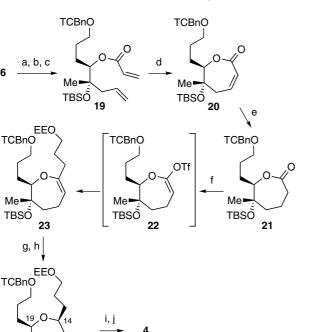


Scheme 3. Reagents and conditions: (a) HNMe(OMe)·HCl, AlMe₃, CH₂Cl₂, 0 °C, 3 h; (b) NaH, DMF, 2,4,6-trichlorobenzyl bromide, $0 \rightarrow 23$ °C, 4 h, 84%; (c) isopropenyl magnesium bromide, THF, $-78 \rightarrow -20$ °C, 4 h, 98%; (d) NaBH₄, CeCl₃·7H₂O, MeOH, 22 °C, 1 h, 98%; (e) (-)-DIPT, Ti(O'Pr)₄, TBHP, CH₂Cl₂, -25 °C, 12 h, (-)-17: 47% (97% ee), 18: 44% (88% ee); (f) PhCO₂H, DEAD, Ph₃P, THF, 23 °C, 6.5 h, 77% (92% ee); (g) DIBAH, CH₂Cl₂, -78 °C, 30 min, 98%; (h) VO(acac)₂, TBHP, PhH, 23 °C, 6.5 h, 80%; (i) NaH, BnBr, Bu₄NI, THF, $0 \rightarrow 23$ °C, 17 h, 100%; (j) Sn(CH=CH₂)₄, BuLi, THF, -78 °C, 1 h, then CuCN, -78 °C, 10 min, then BF₃·OEt₂, -78 °C, 10 min, then benzyl ether of 18, $-78 \rightarrow 0$ °C, 80 min, 99%.

afford 14 in 93% yield for three steps. The TBS group of 14 was removed and the resulting alcohol was treated with Tf₂O to give 15 (99%), which was converted to 5 by a substitution reaction with lithiated methyl methylthiomethyl sulfoxide in 96% yield. Thus, the A-ring segment 5 was obtained in 14 steps in 20% overall yield from 8.

In the preparation of 6, depicted in Scheme 3, we employed Sharpless asymmetric epoxidation for kinetic resolution of 17 in order to construct the stereocenters at C18 and C19 of 6.¹⁰ The allyl alcohol 17 was provided by a four-step process from 7 [Weinreb amidation,¹¹ 2,4,6-trichlorobenzyl (TCBn) protection, addition of isopropenyl magnesium bromide, and Luche reduction]¹² in 81% overall yield. When 17 was treated with (-)-diisopropyl tartrate and *tert*-butyl hydroperoxide in the presence of $Ti(O'Pr)_4$ at $-25 \circ C$, the desired epoxide 18 was obtained in 88% ee (44% yield) along with recovered (-)-17 as an optically active form in 97% ee (47% yield).¹³ The antipodal (-)-17 was also convertible to 18 through a three-step process including Mitsunobu reaction,¹⁴ benzoyl deprotection, and diastereoselective epoxidation¹⁵ (92% ee and 60% overall yield).^{13,16} The optically active 18¹⁷ was protected as a benzyl ether and treated with a higher-order divinylcyanocuprate in the presence of $BF_3 \cdot OEt_2^{18}$ to give 6 in 99% yield for two steps.

Construction of the D-ring 4 from 6 is shown in Scheme 4. Transformation of 6 into 19 (87% overall yield) through a three-step process [protection of the tertiary alcohol with a TBS group, removal of the benzyl group, and formation of an acryloyl ester] followed by a ringclosing olefin metathesis reaction using the second-generation Grubbs' catalyst¹⁹ produced α , β -unsaturated lactone 20 (89% from 19), which was reduced to 21 in quantitative yield. The lactone 21 was converted to cyclic vinyl triflate 22, which was treated in situ with a



6

Me

TBSO

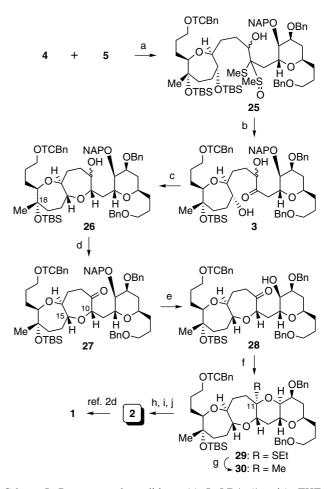
24

ÓTBS

Scheme 4. Reagents and conditions: (a) TBSOTf, 2,6-lutidine, CH_2Cl_2 , $0 \rightarrow 23 °C$, 9 h, 99%; (b) DDQ, $CH_2Cl_2-H_2O$ (10:1), 23 °C, 14.5 h, 93%; (c) acryloyl chloride, ^{*i*}Pr₂NEt, CH₂Cl₂, $0 \rightarrow 25$ °C, 4 h, 94%; (d) [(MesNCH₂)₂C](Cy₃P)Cl₂Ru=CHPh (cat.), (CH₂Cl)₂, reflux, 14 h, 89%; (e) H₂, 5% Pt/C, EtOAc, 23 °C, 5 h, 100%; (f) KHMDS, HMPA, THF, -78 °C, 2 h, then PhNTf₂, -78 °C, 1 h, then EE- $OCH_2CH_2CH_2MgBr$, CuI, Me₂S, $-78 \rightarrow -40$ °C, 13.5 h; (g) thexylborane, THF, 0 °C, 13.5 h, then 3 M NaOH, TBHP, $0 \rightarrow 23$ °C, 1.5 h, 65% from **21**; (h) TBSOTf, 2,6-lutidine, CH_2Cl_2 , $0 \rightarrow 23$ °C, 6 h, 100%; (i) PPTS (cat.), MeOH, 23 °C, 8 h, 87%; (j) (COCl)₂, DMSO, CH₂Cl₂, $-78 \,^{\circ}\text{C}$, 15 min, then Et₃N, $-78 \rightarrow 0 \,^{\circ}\text{C}$, 30 min, 100%.

Grignard reagent, prepared from protected 3-bromopropanol, in the presence of copper iodide to afford cyclic vinyl ether 23.²⁰ Hydroboration of 23 with thexylborane and the subsequent oxidation provided 24 stereoselectively in 65% overall yield from 21. Deprotection followed by Swern oxidation⁸ gave 4 in 87% yield for two steps.¹⁷ The D-ring 4 was, thus, synthesized in 20 steps including the stereochemical inversion process in 20% overall yield from 7.

Assembly of the ABCD-ring part is outlined in Scheme 5. Deprotonation of 5 with LDA followed by the reaction with 4 (0.37 equiv) produced 25 as a mixture of diastereomers (86% yield from 4, 57% recovery of 5). The TBS and methylthio methysulfinyl acetal groups of 25 were removed under less acidic conditions to give 3 (71% yield), which was cyclized by short-time (within 10 min) treatment with Et₃SiH in the presence of TMSOTf at -20 °C to produce **26** as a mixture of two diastereomers in 65% yield.^{3,4,17,21} The alcohol **26** was oxidized with Dess-Martin periodinane²² to give the desired 27 as a single stereoisomer (100% yield), the stereochemistry of which was confirmed by the presence of NOE between H10 and H15. The NAP group of 27 was removed with DDQ to afford 28 in 93% yield.⁹



Scheme 5. Reagents and conditions: (a) 5, LDA (1 equiv), THF, -78 °C, 15 min, then 4 (0.37 equiv), -78 °C, 15 min, 86%; (b) TFA-THF-H₂O (1:5:5), $0 \rightarrow 23 \text{ °C}$, 7.5 h, 71%; (c) TMSOTf, Et₃SiH, MS4A, CH₂Cl₂, -20 °C, 10 min, 65%; (d) DMPI, CH₂Cl₂, 25 °C, 1 h, 100%; (e) DDQ, CH₂Cl₂–MeOH–H₂O (40:10:1), $0 \rightarrow 25 \,^{\circ}$ C, 1 h, 93%; (f) EtSH, $Zn(OTf)_2$, CH_2Cl_2 , $0 \rightarrow 23 \,^{\circ}C$, 6.5 h, 85% (after three cycles); (g) mCPBA, CH₂Cl₂, 0 °C, 30 min, then AlMe₃, 0 °C, 1 h, 92%; (h) DDQ, CH₂Cl₂-H₂O (10:1), 25 °C, 5 h, 65%; (i) TIPSOTf, 2,6-lutidine, DMF, 70 °C, 5 h; (j) H₂, 10% Pd/C, EtOH, 25 °C, 2 h, 83% (two steps).

Reaction of **28** with EtSH and $Zn(OTf)_2$ in CH_2Cl_2 did not complete even after 27 h and gave cyclic S,O-acetal **29** in 56% yield along with recovered **28** (42%); therefore, the process was repeated twice in order to obtain a reasonable amount of **29** (85% yield after three cycles).²³ Introduction of an angular methyl group at C11 was performed by oxidation of 29 with mCPBA followed by in situ treatment with AlMe₃ according to the Nicolaou procedure to produce 30 in 92% yield.⁵ The tetracyclic 30 was converted to 2 { $[\alpha]_D^{25}$ +23.6 (CHCl₃, c 0.29)} in 54% total yield through a three-step process including selective debenzylation, TIPS protection, and TCBn deprotection. The spectral data (¹H and ¹³C NMR as well as IR) and optical rotation of the resulting 2 agreed well with those of the literature.^{2d,h} Thus, the A- and Dring segments (5 and 4) were convergently assembled into the ABCD-ring part 2 (16% overall yield for 10 steps from 4), which was already converted to 1 by the Yamamoto group,^{2d} thereby completing the formal total synthesis of 1.

In conclusion, the concise construction of the *trans*fused 7/7/6/6 tetracyclic ether (ABCD-ring) part 2 of hemibrevetoxin B (1) from the A- and D-ring segments (5 and 4) was achieved by a convergent process including coupling reaction of the acyl anion equivalent 5 with 4, reductive cyclization of α , ε -dihydroxyketone 3, and introduction of a methyl group at the central ring junction of 29 by the Nicolaou method.

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References and notes

- Prasad, A. V. K.; Shimizu, Y. J. Am. Chem. Soc. 1989, 111, 6476–6477.
- 2. (a) Nicolaou, K. C.; Reddy, K. R.; Skokotas, G.; Sato, F.; Xiao, X.-Y. J. Am. Chem. Soc. 1992, 114, 7935-7936; (b) Nicolaou, K. C.; Reddy, K. R.; Skokotas, G.; Sato, F.; Xiao, X.-Y.; Hwang, C.-K. J. Am. Chem. Soc. 1993, 115, 3558-3575; (c) Kadota, I.; Park, J.-Y.; Koumura, N.; Pollaud, G.; Mitsukawa, Y.; Yamamoto, Y. Tetrahedron Lett. 1995, 36, 5777-5780; (d) Kadota, I.; Yamamoto, Y. J. Org. Chem. 1998, 63, 6597-6606; (e) Nakata, T.; Nomura, S.; Matsukura, H.; Morimoto, M. Tetrahedron Lett. 1996, 37, 217-220; (f) Morimoto, M.; Matsukura, H.; Nakata, T. Tetrahedron Lett. 1996, 37, 6365-6368; (g) Mori, Y.; Yaegashi, K.; Furukawa, H. J. Am. Chem. Soc. 1997, 119, 4557-4558; (h) Mori, Y.; Yaegashi, K.; Furukawa, H. J. Org. Chem. 1998, 63, 6200-6209; (i) Rainier, J. D.; Allwein, S. P.; Cox, J. M. Org. Lett. 2000, 2, 231-234; (j) Rainier, J. D.; Allwein, S. P.; Cox, J. M. J. Org. Chem. 2001, 66, 1380-1386; (k) Zakarian, A.; Batch, A.; Holton, R. A. J. Am. Chem. Soc. 2003, 125, 7822-7824; see also, (1) Feng, F.; Murai, A. Chem. Lett. 1992, 1587-1590; (m) Feng, F.; Murai, A. Chem. Lett. 1995, 23-25; (n) Feng, F.; Murai, A. Synlett 1995, 863-865; (o) Ishihara, J.; Murai, A. Synlett 1996, 363–365.
- 3. Nicolaou, K. C.; Hwang, C.-K.; Nugiel, D. A. J. Am. Chem. Soc. 1989, 111, 4136–4137.
- (a) Fujiwara, K.; Saka, K.; Takaoka, D.; Murai, A. Synlett 1999, 1037–1040; (b) Fujiwara, K.; Takaoka, D.; Kusumi, K.; Kawai, K.; Murai, A. Synlett 2001, 691–693;

(c) Fujiwara, K.; Koyama, Y.; Kawai, K.; Tanaka, H.; Murai, A. *Synlett* **2002**, 1835–1838.

- Nicolaou, K. C.; Prasad, C. V. C.; Hwang, C.-K.; Duggan, M. E.; Veale, C. A. J. Am. Chem. Soc. 1989, 111, 5321–5330.
- (a) Ogura, K.; Tsuchihashi, G. *Tetrahedron Lett.* **1972**, *13*, 2681–2684;
 (b) Herrmann, J. L.; Richman, J. E.; Wepplo, P. J.; Schlessinger, R. H. *Tetrahedron Lett.* **1973**, *14*, 4707–4710.
- Yadav, J. S.; Reddy, B. V. S.; Reddy, K. B.; Satyanarayana, M. *Tetrahedron Lett.* 2002, 43, 7009–7012.
- Mancuso, A. J.; Huang, S.-L.; Swern, D. J. Org. Chem. 1978, 43, 2048.
- (a) Gaunt, M. J.; Yu, J.; Spencer, J. B. J. Org. Chem. 1998, 63, 4172–4173; (b) Xia, J.; Abbas, S. A.; Locke, R. D.; Piskorz, C. F.; Alderfer, J. L.; Matta, K. L. Tetrahedron Lett. 2000, 41, 169–173; (c) Wright, J. A.; Yu, J.; Spencer, J. B. Tetrahedron Lett. 2001, 42, 4033–4036.
- (a) Martin, V. S.; Woodard, S. S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K. B. J. Am. Chem. Soc. 1981, 103, 6237–6240; (b) Gao, Y.; Klunder, J. M.; Hanson, R. M.; Masamune, H.; Ko, S. Y.; Sharpless, K. B. J. Am. Chem. Soc. 1987, 109, 5765–5780.
- (a) Basha, A.; Lipton, M.; Weinreb, S. M. *Tetrahedron Lett.* **1977**, 4171–4174; (b) Levin, J. I.; Turos, E.; Weinreb, S. M. *Synth. Commun.* **1982**, *12*, 989–993.
- 12. Gemal, A. L.; Luche, J.-L. J. Am. Chem. Soc. 1981, 103, 5454–5459.
- Enantiomeric purity of 17 and 18 was determined by HPLC analysis using chiral columns (CHIRALCEL AD [DAICEL] for 18, CHIRALCEL OD-H [DAICEL] for benzoate ester of 17).
- 14. Mitsunobu, O. Synthesis 1981, 1-28.
- 15. Rossiter, B. E.; Verhoven, T. R.; Sharpless, K. B. *Tetrahedron Lett.* **1979**, *20*, 4733–4736.
- 16. Slight lowering of the enantiomeric purity was observed in the Mitsunobu reaction step.
- 17. Since epoxide 18 was used as it stood (88–92% ee), the optical purity of 4 was same as that of 18. Although the minor enantiomer of 4 produced small amounts of undesired diastereomers, which were inseparable from 25, in the connection reaction with 5, these were facilely separated from 26 at the C-ring cyclization step.
- 18. Alexakis, A.; Jachiet, D.; Normant, J. F. *Tetrahedron* **1986**, *42*, 5607–5619.
- Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. Org. Lett. 1999, 1, 953–956.
- (a) Tsushima, K.; Araki, K.; Murai, A. Chem. Lett. 1989, 1313; (b) Tsushima, K.; Murai, A. Chem. Lett. 1990, 761.
- 21. Prolonged reaction time resulted in elimination of the TBSO group at C18 of **26** to produce a significant amount of the corresponding C18-exomethylene derivative.
- (a) Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4155;
 (b) Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. 1991, 113, 7277.
- 23. The cyclic *S*,*O*-acetal formation reaction of **28** usually stopped at about 50% conversion after 6h. Several attempts to improve the yield (changing solvent to MeCN or MeNO₂, heating, or changing concentration of **28** and reagents) showed no effect.